

This article was downloaded by:

On: 29 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713618290>

"DRY MEDIA PROCESS": AN EFFICIENT TECHNIQUE FOR FAST REGIOSPECIFIC SYNTHESSES OF SPIRANIC CYCLOPHOSPHAZENE CRYPTANDS

Sylvie Scheidecker^a; Delphine Semenzin^b; Guita Etemad-moghadam^b; Francois Sournies^a; Max Koenig^b; Jean-Francois Labarre^a

^a CNRS, Laboratoire Structure et Vie-Université Paul Sabatier-118, Toulouse Cedex, France ^b

Laboratoire d'Activation Moléculaire par l'Electricité, le Rayonnement et l'Energie Sonore (AMPERES), Université Paul Sabatier-118, Toulouse Cedex, France

To cite this Article Scheidecker, Sylvie , Semenzin, Delphine , Etemad-moghadam, Guita , Sournies, Francois , Koenig, Max and Labarre, Jean-Francois(1993) "“DRY MEDIA PROCESS”: AN EFFICIENT TECHNIQUE FOR FAST REGIOSPECIFIC SYNTHESSES OF SPIRANIC CYCLOPHOSPHAZENE CRYPTANDS', *Phosphorus, Sulfur, and Silicon and the Related Elements*, 80: 1, 85 — 88

To link to this Article: DOI: 10.1080/10426509308036880

URL: <http://dx.doi.org/10.1080/10426509308036880>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

“DRY MEDIA PROCESS”: AN EFFICIENT TECHNIQUE FOR FAST REGIOSPECIFIC SYNTHESES OF SPIRANIC CYCLOPHOSPHAZENE CRYPTANDS

SYLVIE SCHEIDECKER,[†] DELPHINE SEMENZIN,[‡]
GUITA ETEMAD-MOGHADAM,[‡] FRANCOIS SOURNIES,[†]
MAX KOENIG*[‡] and JEAN-FRANCOIS LABARRE*[†]

[†]CNRS, Laboratoire Structure et Vie-Université Paul Sabatier-118, route de
Narbonne-31062 Toulouse Cedex, France; and [‡]Laboratoire d'Activation
Moléculaire par l'Electricité, le Rayonnement et l'Energie Sonore (AMPERES)-
Université Paul Sabatier-118, route de Narbonne-31062 Toulouse Cedex, France

(Received January 28, 1993; in final form March 4, 1993)

The aminolysis of the hexachlorocyclotriphosphazene, $N_3P_3Cl_6$ **1**, by the oxodiamines $H_2N-(CH_2)_3-O-(CH_2)_2-O-(CH_2)_3-NH_2$ **2** and $H_2N-(CH_2)_3-O-(CH_2)_6-O-(CH_2)_3-NH_2$ **3** occurs immediately and regio-specifically in dry media leading to the corresponding spiranic cryptands **4** and **5**, respectively. This process is an actual improvement (time, selectivity and workup) on the standard reactions.

Key words: Cyclotriphosphazene; regiospecific aminolysis; dry media; alumina supported reaction.

Reactions of long oxodiamines, $H_2N-(CH_2)_n-O-(CH_2)_m-O-(CH_2)_n-NH_2$, with hexachlorocyclotriphosphazene ($N_3P_3Cl_6$) were investigated since 1989 with the aim of designing highly selective antitumoral drugs with acceptable solubility in physiological serum.^{1,2} These reactions lead to macrocyclic host molecules, the conformation, the cavity size and the number of coordination sites of which depend drastically on experimental conditions.^{3–11} Most of these attractive one- and two-ring architectures were unambiguously evidenced by x-ray investigations^{6–11} and only four configurations were ever observed up to now (Figure 1):

- (i) the SPIRO configuration (in which the oxodiamino ligand is grafted as a SPIRO loop onto one phosphorus atom of one N_3P_3 ring),
- (ii) the cis-ANSA configuration (in which the oxodiamino ligand is grafted on one side of one N_3P_3 ring as an ANSA arch onto two different phosphorus atoms),
- (iii) the trans-ANSA configuration (in which the oxodiamino ligand is grafted on both sides of one N_3P_3 ring as an ANSA arch onto two different phosphorus atoms) and
- (iv) the BINO configuration (in which the oxodiamino ligand bridges two different N_3P_3 rings).

*Authors for correspondence.

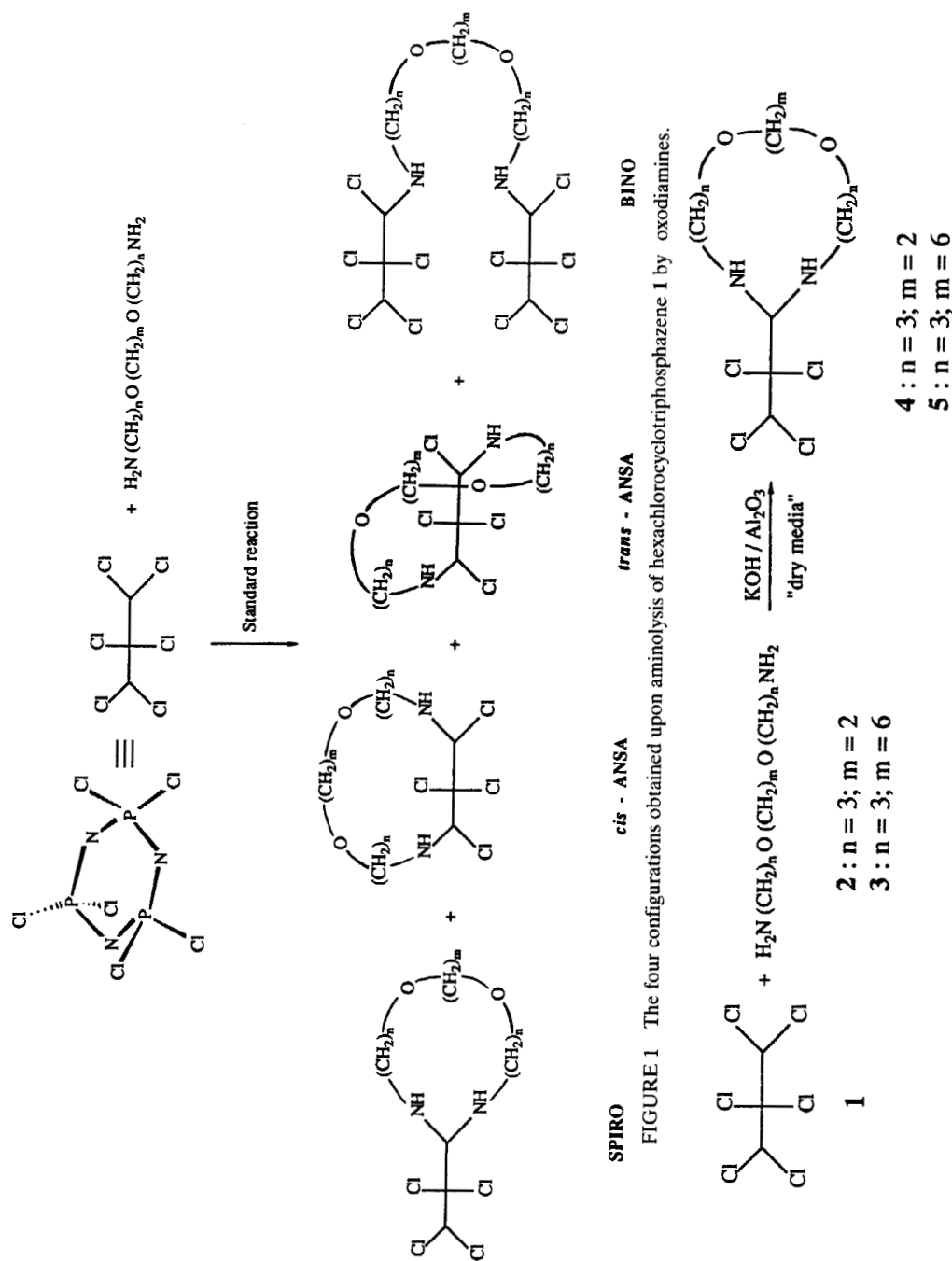


FIGURE 2 Regiospecific synthesis of 4-5 in "dry media."

However, such syntheses are not, strictly speaking, regioselective, every configuration being currently obtained with some others as by-products, and they need 48 hours at least to be achieved when performed at room temperature.

According to the capacity of these macrocyclic host cyclophosphazenes to bind metallic moieties and to yield so new cryptates for further applications,¹² it was of interest to attempt at the production of such "cages" in a more regioselective and less time-and-money consuming way. The present contribution reports on the improvements we got by using "dry media techniques".

It is well-known indeed that a wide variety of chemical reactions can be promoted in heterogeneous media thanks to the acidic and/or basic sites located on the surface of suitable solids such as alumina, clay, silica gel, talc and others.¹³ The significant advantages of such "dry media" reactions relatively to the corresponding homogeneous reactions are the milder conditions, the more specific (chemospecific, regiospecific and stereospecific) transformations and the easier isolation of final pure products.

Syntheses of SPIRO cryptands from oxodiamines were extensively reported²: they were carried out in an heterogeneous liquid/liquid medium (saturated aq. Na_2CO_3 /toluene) at room temperature for 48 hours. The completion of the reactions was followed by ^{31}P NMR spectroscopy and the percentage of each component in the final crude product was estimated by NMR. Thus, starting from **2**, the latter contains the expected SPIRO moiety **4** (80%) as the major product with 5% of BINO, 10% of unidentified by-products and 5% of unreacted $\text{N}_3\text{P}_3\text{Cl}_6$ **1**. Incidentally, the same situation occurs for most of other oxodiamines, except for **3** where the corresponding SPIRO **5** is no more obtained as the major product (5% only), its trans-ANSA isomer being here the main component of the crude final mixture (50%) with 10% of BINO, 30% of unidentified by-products and 5% of unreacted $\text{N}_3\text{P}_3\text{Cl}_6$ **1**. These reactions occur slowly on interface between the two liquid systems and require the use of important quantities of solvents (toxicity and workup problems).

These syntheses were repeated on impregnated alumina or talc in the absence of solvent ("dry media process") (Figure 2) and the desired SPIRO derivatives [**4** as well as **5**] were so obtained in $\geq 90\%$ yield (see experimental section).

Many solid mineral supports (neutral or activated basic alumina, lamellar structures such as talcs) were tested. The influence of the nature of the basic system (potassium fluoride, potassium hydroxide, sodium carbonate or oxodiamine supported on the inorganic solid) was examined. Commercial basic alumina or sodium carbonate are not efficient conversely to the more basic potassium hydroxide supported on alumina. The amount of water is a critical parameter since the selectivity of the reaction depends drastically on the degree of dryness of basic supports (for example, 90% **5** were obtained from the alumina-supported KOH when dried at 70°C for 24 h in oven).

Acetone as the eluent has to be definitely preferred concerning the regioselectivity of the reaction. Indeed, a decrease of this selectivity is observed when elution is performed with other polar solvents such as acetonitrile.

Then, the use of alumina or alumina-supported KOH as mineral support and acetone as eluent makes syntheses of SPIRO cyclophosphazenic cryptands fast and regiospecific. Other inorganic supports will be tested with the aim of obtaining cis-

ANSA, trans-ANSA and BINO entities in such a time-and-money non-consuming way.

EXPERIMENTAL

The NMR spectra were recorded on a Bruker AC 200 spectrometer with H_3PO_4 85% as external reference.

Hexachlorocyclotriphosphazene **1** was generously provided to us (degree of purity $\geq 99.8\%$) by SHIN NISSO KAKO Co, subsidiary company of NIPPON SODA Co. BASF and TEXACO supplied us with the oxodiamines **2** and **3** (degree of purity $\geq 98\%$).

Alumina-supported potassium hydroxide: Potassium hydroxide (Prolabo Rectapur, 11 g) in H_2O (250 ml) was mixed with neutral chromatographic alumina (Fluka, type 207 C, 90–110 μ , 50 g). After stirring for 5 min, the water was removed under reduced pressure. The resulting powder was further dried at 70°C for 24 h in oven. This reagent may be kept in a dessicator without loss of activity during several weeks.

Talc-supported sodium carbonate: Sodium carbonate (Prolabo, 27 g) is dissolved in H_2O (250 ml) and talc (Prolabo Rectapur, 50 g) is added under stirring. The water is removed in vacuo. The impregnated talc is then dried in a drying oven (80°C—2 days) and stored without loss of activity during several weeks.

General method for the preparation of spiranic cyclophosphazene cryptands **4 and **5**:** A mixture of **1** (0.40 g; 1.1 mmol) and the dioxodiamine **2** or **3** (1 eq.; 1.1 mmol) in toluene (20 ml) was added to solid support ($\text{Na}_2\text{CO}_3/\text{talc}$ or $\text{KOH}/\text{alumina}$) (3.5 g. corresponding to 10 eq. of base). The solvent was immediately removed under reduced pressure at room temperature and the resulting powder was eluted with an organic solvent. The unreacted **1** was extracted with *n*-hexane and a second extraction with acetone gives the spiranic compounds **4** and **5** as pure form in $\geq 90\%$ yield. The analytical parameters of **4–5** are in good agreement with the literature data.^{2,10–11}

Remark: The stoichiometry of **1/2** or **3** was modified from 1/1 to 1/2 if neutral alumina was used as solid support, the excess of the oxodiamine operating as a base.

ACKNOWLEDGEMENTS

The authors are greatly indebted to the Paul Sabatier University for its generous financial support to this work through a 1991–1992 ACRU dotation.

REFERENCES

1. C. W. Allen, *Chem. Rev.*, **91**, 119 (1991).
2. (a) A. El Bakili, P. Castera, J-P. Faucher, F. Sournies and J-F. Labarre, *J. Mol. Struct.*, **195**, 21 (1989); (b) A. El Bakili, P. Castera, J-P. Faucher, F. Sournies, J-F. Labarre, R. Enjalbert and J. Galy, *ibid*, **196**, 207 (1989); (c) J-F. Labarre, *Advances in Supramolecular Chemistry*, in press.
3. F. Sournies, P. Castera, A. El Bakili and J-F. Labarre, *J. Mol. Struct.*, **221**, 239 (1990).
4. F. Sournies, P. Castera, A. El Bakili, J-P. Faucher, M. Graffeuil and J-F. Labarre, *J. Mol. Struct.*, **221**, 245 (1990).
5. F. Sournies, A. El Bakili, J-F. Labarre and B. Perly, *J. Mol. Struct.*, **196**, 201 (1989).
6. F. Sournies, A. El Bakili, B. Zanin, J-F. Labarre and J. Jaud, *J. Mol. Struct.*, **220**, 43 (1990).
7. T. S. Cameron, A. Linden, A. El Bakili, P. Castera, J-P. Faucher, M. Graffeuil, F. Sournies and J-F. Labarre, *J. Mol. Struct.*, **212**, 281 (1989).
8. J. Jaud, F. Sournies and J-F. Labarre, *J. Mol. Struct.*, **212**, 305 (1989).
9. R. Enjalbert, J. Galy, F. Sournies and J-F. Labarre, *J. Mol. Struct.*, **221**, 253 (1990).
10. B. Zanin, F. Sournies, J-F. Labarre, R. Enjalbert and J. Galy, *J. Mol. Struct.*, **240**, 77 (1990).
11. B. Zanin, S. Scheidecker, F. Sournies and J-F. Labarre, *J. Mol. Struct.*, **246**, 133 (1991).
12. M. Veith, M. Kross and J-F. Labarre, *J. Mol. Struct.*, **243**, 189 (1991) and **248**, 345 (1991).
13. Cf. for example: (a) A. Foucaud, G. Bram and A. Loupy, in "Preparative Chemistry using Supported Reagents", P. Lazlo Ed., Academic Press, Chap. **17**, 317 (1987). (b) G. Bram and A. Loupy, *ibid*, Chap. **20**, 387 (1987). (c) R. Latouche, F. Texier-Boullet and J. Hamelin, *Tetrahedron Lett.*, **32**, 1179 (1991). (d) A. Loupy, G. Bram and J. Sansoulet, *New J. Chem.*, **16**, 233 (1992).